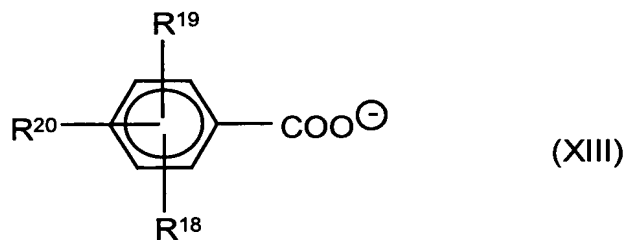
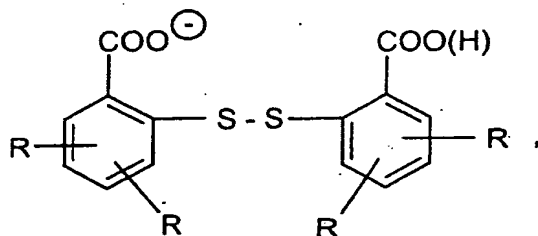
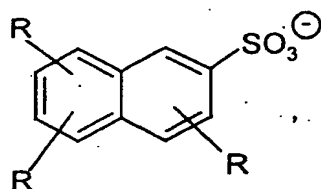
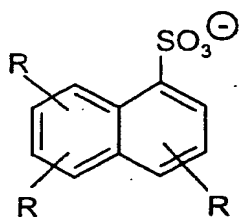
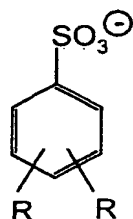
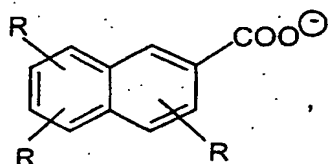
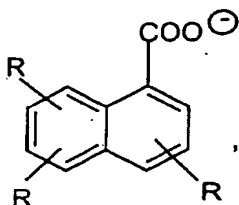


# Amendments to the Claims

1. (Currently Amended) ~~The use of layered double hydroxide salts as charge control agents in~~ A process for controlling the charge of an electrophotographic toners and developers, in toner, electrophotographic developer, powder, powder coating materials ~~material, electret materials~~ material or a chargeable material and in an electrostatic separation processes of chargeable materials process comprising the step of adding at least one charge control agent to the electrophotographic toner, electrophotographic developer, powder, powder coating material, electret material or the chargeable material in an electrostatic separation process, wherein the at least one charge control agent is a layered double hydroxide salt and wherein the double hydroxide salt contains ~~comprises~~ at least one of monovalent and/or and divalent metal cations and also ~~trivalent metal cations, and also contains~~ organic anions A of the formulae below ~~(XIII)~~



~~in which~~ wherein  $R^{18}$ ,  $R^{19}$  and  $R^{20}$  are identical or different and are hydrogen,  $C_1$ - $C_{22}$ -alkyl,  $C_1$ - $C_{18}$ -alkenyl,  $C_1$ - $C_{18}$ -alkoxy;



in which wherein R is hydrogen,  $\text{C}_1\text{-C}_4\text{-alkyl}$ ,  $\text{C}_1\text{-C}_4\text{-alkoxy}$  or halogen.

2. (Currently Amended) The use-process as claimed in claim 1, wherein as ~~organic anion~~ the organic anions are anions from an acid selected from the group consisting of an anion of the following acids is used:  
benzoic acid, naphthoic acid, 4-tert-butylbenzoic acid, benzenesulfonic acid, p-toluenesulfonic acid, naphthalenesulfonic acid, and 2,2'-dithiobenzoic acid.
3. (Currently Amended) The use-process as claimed in claim 1 ~~or 2~~, wherein the number of hydroxyl groups is from about 1.8 to 2.2 times the sum of all the metal cations.
4. (Currently Amended) The use-process as claimed in ~~at least one of claims 1 to 3~~ claim 1, wherein monovalent metal cations ~~present are those selected~~ from the group consisting of  $\text{Li}^+$ ,  $\text{Na}^+$  and  $\text{K}^+$ , divalent metal cations ~~present are those from the group~~  $\text{Mg}^{2+}$ ,  $\text{Ca}^{2+}$ ,  $\text{Zn}^{2+}$ ,  $\text{Co}^{2+}$ ,  $\text{Ni}^{2+}$ ,  $\text{Fe}^{2+}$ ,  $\text{Cu}^{2+}$  and  $\text{Mn}^{2+}$ , and trivalent metal cations ~~present are those from the group~~  $\text{Al}^{3+}$ ,  $\text{Fe}^{3+}$ ,  $\text{Co}^{3+}$ ,  $\text{Mn}^{3+}$ ,  $\text{Ni}^{3+}$ ,  $\text{Cr}^{3+}$  and  $\text{B}^{3+}$ .
5. (Currently Amended) The use-process as claimed in ~~at least one of claims 1 to 4~~ claim 1, wherein the at least one double hydroxide salts ~~salt contains~~ contain  $\text{Mg}^{2+}$  and  $\text{Al}^{3+}$ .
6. (Currently Amended) The use-process as claimed in claim 5, wherein the molar ratio  $\text{Mg}^{2+} : \text{Al}^{3+}$  is from 3.1:1 to 1:2.
7. (Currently Amended) The use-process as claimed in ~~at least one of claims 1 to 6~~ claim 1, wherein the at least one double hydroxide salt is a calcined hydrotalcite.
8. (Currently Amended) The use-process as claimed in ~~at least one of claims 1 to 7~~, claim 1, wherein the adding step further comprises adding ~~wherein the double hydroxide salt is used in combination with one or more further charge control agents selected from the group consisting of~~ triphenylmethanes; ammonium and compounds, immonium compounds, iminium compounds; fluorinated ammonium

~~and compounds~~, fluorinated immonium compounds; biscationic acid amides; polymeric ammonium compounds; diallylammonium compounds; aryl sulfide derivatives, phenol derivatives; phosphonium compounds, ~~and~~ fluorinated phosphonium compounds; calix[n]arenes, cyclically linked oligosaccharides, ~~(cyclodextrins) and their derivatives of cyclically linked oligosaccharides~~, in particular boron ester derivatives, interpolyelectrolyte complexes (IPECs); polyester salts; metal complex compounds, ~~especially salicylate-metal complexes and salicylate-nonmetal complexes~~, salts of ionic structured silicates, hydroxycarboxylic acid-metal complexes ~~and~~ hydroxycarboxylic acid-nonmetal complexes, benzimidazolones; azines, thiazines ~~or and~~ oxazines, ~~which are listed in the Colour Index as Pigments, Solvent Dyes, Basic Dyes or Acid Dyes.~~

9. (Currently Amended) The ~~use process~~ as claimed in ~~at least one of claims 1 to 8~~ claim 1, wherein the at least one charge control agent is present in a concentration of from 0.01% to 50% by weight, based on the total weight of the electrophotographic toner, electrophotographic developer, coating material, powder, powder coating material, electret material or materials chargeable material for electrostatic separation.

10. (Currently Amended) An electrophotographic toner, powder or powder coating material, containing from 30% to 99.99% by weight of a binder, from 0.01% to 50% by weight of at least one layered double hydroxide salt as set forth in ~~claims 1 to 7~~ claim 1, and, ~~if desired~~ optionally, from 0.001% to 50% by weight of a colorant, based ~~in each case~~ on the total weight of the electrophotographic toner, powder or powder coating material.

11. (New) The process as claimed in claim 1, wherein the divalent metal cations are selected from the group consisting of  $Mg^{2+}$ ,  $Ca^{2+}$ ,  $Zn^{2+}$ ,  $Co^{2+}$ ,  $Ni^{2+}$ ,  $Fe^{2+}$ ,  $Cu^{2+}$  and  $Mn^{2+}$ .

12. (New) The process as claimed in claim 1, wherein the trivalent metal cations are selected from the group consisting of  $\text{Al}^{3+}$ ,  $\text{Fe}^{3+}$ ,  $\text{Co}^{3+}$ ,  $\text{Mn}^{3+}$ ,  $\text{Ni}^{3+}$ ,  $\text{Cr}^{3+}$  and  $\text{B}^{3+}$ .
13. (New) The process as claimed in claim 8, wherein the metal complex compound is a salicylate metal complex.
14. (New) An electrophotographic toner, electrophotographic developer, powder, powder coating material, electret material or chargeable material for an electrostatic separation process made in accordance with the process of claim 1.